

Ammonia-Nitrogen Recovery from Synthetic Solution using Agricultural Waste Fibers

A.Y. Zahrim^{1*}, L. N. S. Ricky¹, N. Hilal² and K. F. Tamrin³

¹Department of Chemical Engineering, Universiti Malaysia Sabah, Jalan UMS, 88400 Kota Kinabalu, Sabah, Malaysia; zahrim@ums.edu.my, rickylns@hotmail.com

²Centre for Water Advanced Technologies and Environmental Research (CWATER), College of Engineering, Swansea University, Swansea SA2 8PP, UK; n.hilal@swansea.ac.uk

³Department of Mechanical and Manufacturing Engineering, Faculty of Engineering Universiti Malaysia Sarawak (UNIMAS) Kota Samarahan, Malaysia; k.f.tamrin@outlook.com

Abstract

Background/Objectives: In this study, modification of Empty Fruit Bunch (EFB) fibers as a means to recover ammonia nitrogen from a synthetic solution was investigated. **Methods:** The EFB fiber was modified using sodium hydroxide. Adsorption-desorption studies of ammonia nitrogen into the modified EFB fiber were investigated. **Findings:** The increase in adsorption capacity was found to be proportional with the increase of pH up to 7, temperature and ammonia concentration. The maximum adsorption capacity is 0.53-10.89 mg/g. The attachment of ammonia nitrogen involves ion exchange-chemisorption. The maximum desorption capacity of 0.0999 mg/g. **Applications:** This study can be used as a baseline for designing a low cost adsorbent system for ammonia nitrogen recovery drainage and industrial wastewater as well as EFBs-palm oil mill effluent composting.

Keywords: Ammonia Nitrogen, Agricultural Waste, Desorption, Empty Fruit Bunch, Nutrient Recovery

1. Introduction

Industrial nitrogen can be discharged in large volume from pulp and paper, fertilizer, and mining industry¹. Highest nitrogen discharges in pulp and paper industry is due to pulping and bleaching process². Besides that, agricultural drainage and municipal waste are also among the main sources of polluter¹. Ammonia nitrogen concentration greater than 10 mg/L causes intensifying of genotoxicity³. In addition, nitrogen pollution in waterways results in the eutrophication and fouling of rivers, lakes, water reservoirs and oceans. Recovering ammonia nitrogen from polluted water could be an option in treating the contaminated water and simultaneously recycle the nutrient back for agricultural purposes. Of numerous techniques investigated for ammonia nitrogen recovery⁴, a considerable amount of approaches seem to concentrate on developing cheaper and effective agricultural waste

adsorbents⁵. This method is considered eco-friendly, economical, and practically simple to operate⁶.

In Malaysia, Empty Fruit Bunch (EFB) fibers are abundantly available waste with about 91.2 million tons were produced annually⁷. Previously, EFB fibers compost has shown potential in removing ammonia nitrogen from synthetic solution⁸. Degraded fibers could enhance the sorption of ammonia nitrogen due to increasing in negatively charge surface site. However, EFB biodegradation take a long time to biodegrade. To reduce the modification time, chemical modification is suggested in this study. Modification of pine cone powder using sodium hydroxide was found to increase the ammonia nitrogen sorption capacity to 6.15 mg/g⁶. In another study, adsorption capacity of banana peels treated with sodium hydroxide was found to increase from 8.6 to 20.0 mg/g⁹. However, the common purpose for EFB modification studies reported so far is mainly for the productions

*Author for correspondence